Use of stone dust for the production of low CO_2 cements

Shashank Bishnoi, Indian Institute of Technology Delhi



Use of marble and kota stone dust in limestone calcined clay cement (LC³)

Construction and Building Materials 164 (2018) 265-274



Contents lists available at ScienceDirect

Construction and Building Materials

journal homepage: www.elsevier.com/locate/conbuildmat

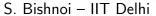
Hydration kinetics and mechanisms of carbonates from stone wastes in ternary blends with calcined clay



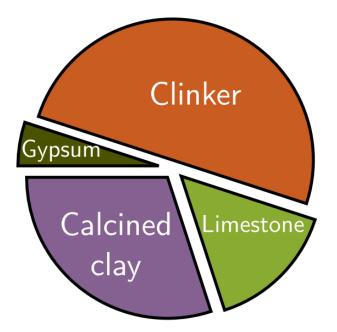
hi

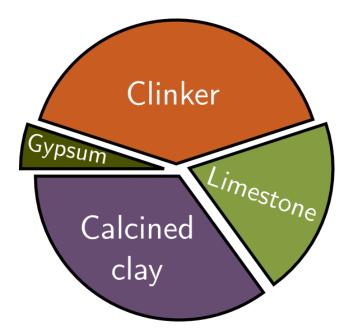
Sreejith Krishnan^{a,*}, Swadesh Kumar Kanaujia^a, Santanu Mithia^b, Shashank Bishnoi^a

^a Department of Civil Engineering, Indian Institute of Technology Delhi, New Delhi 110016, India ^b Development Alternatives, New Delhi 110016, India





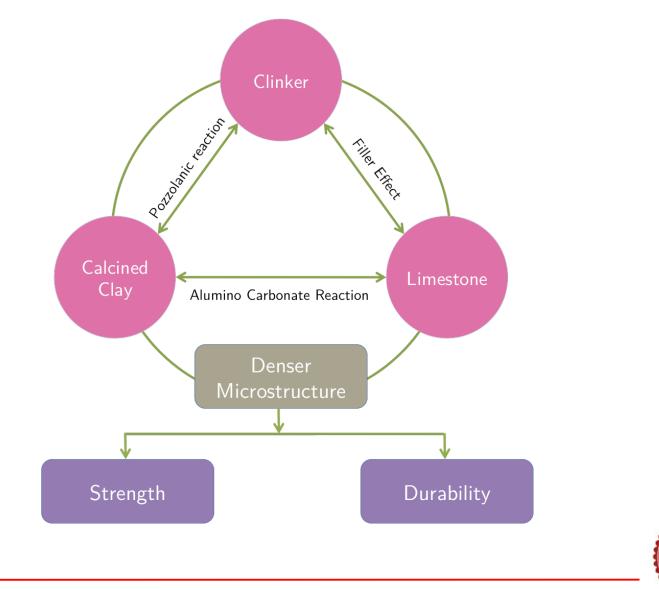






S. Bishnoi – IIT Delhi

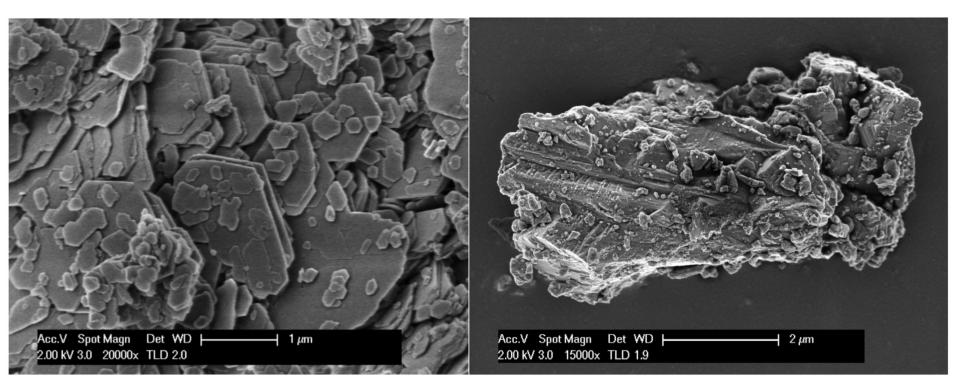
Synergy



S. Bishnoi – IIT Delhi

4

Physical influence



Calcined clay

Limestone



S. Bishnoi – IIT Delhi

What do clay and limestone do?

Calcined clay is known to have a pozzolanic reaction

- Limestone acts as filler
 - Improves rheology,
 - Finer microstructure
- Calcined clay + Limestone:
 - $(Al_2O_3) : 2 \cdot (SiO_2) + CaCO_3$
 - $AI_2O_3 + CaCO_3 + Ca^{2+} + OH^- \Rightarrow C_4ACH_{11}$



The strength of LC³

- Uses raw materials familiar to cement industry
- Raw materials are widely and cheaply available
- Uses available technology commonly used in cement industry
- Use of LC³ is similar to use of conventional cements
- Lower production cost than other cements in most scenarios



The benefits of LC³

- Up to 30% lower CO₂ emissions than other cements
- Lower energy consumption in production than other cements
- Faster strength development than other blended cements
- Excellent durability compared to other cements

Economical, strong and durable infrastructure!



Current understanding: Clay

- Large deposits of clays available in India
 - Required kaolinite content: 40% to 60%
 - Iron content not an issue
- Calcination
 - Dehydroxyllation of kaolinite, completed by 800°C
 - Mullite should not form
 - Petcoke can be used as fuel
- Fineness
 - Very fine clay increases water demand
- Characterisation using TGA & XRD



Do we have clays?

- Pure kaolinitic clays are expensive
 - Ceramics
 - Paint
 - Paper
- They are less available
- Coloured clays are not accepted by the industry
- Cement industry is used to high-silica clays!



Rejected clay from a clay mine





Current understanding: Limestone

- Composition: down to 35% CaO content
 - Dolomitic limestone works
 - Siliceous limestone works
 - Stone wastes available
- Improves workability
- Characterisation using TGA & XRD



Reject limestone from cement plant





Low grade limestone & stone waste











Compositions

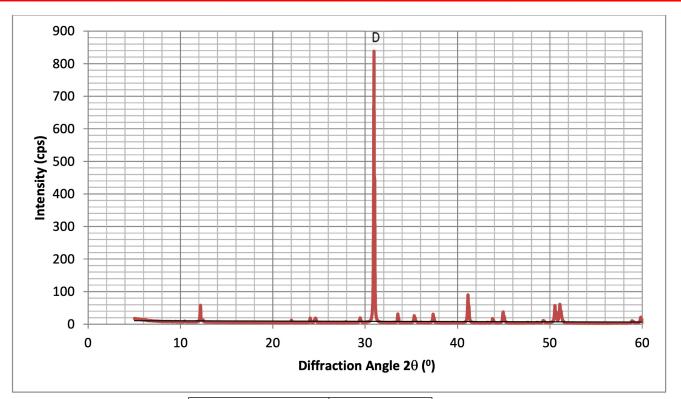
Table 1

Chemical and major phase composition and D_{50} values of the raw materials.

Oxide (%)	OPC	Calcined Clay	BA	KG	KS
SiO ₂	22.07	55.11	4.54	7.45	21.96
Fe ₂ O ₃	4.69	4.18	1.41	0.47	0.76
Al_2O_3	3.75	39.89	0.29	0.62	2.98
CaO	63.35	0.09	29.05	42.29	39.40
MgO	1.08	0.02	24.73	5.44	0.67
SO ₃	2.09	0.104	0.76	0.10	-
Na ₂ O	0.32	0.18	0.08	3.57	0.90
K ₂ O	0.71	0.19	0.08	0.18	0.67
LOI	1.14	-	39.03	39.32	32.94
C ₃ S	56.56	_	_	_	_
C ₂ S	23.83	-	-	_	-
C ₃ A	2.78	-	_	_	-
C ₄ AF	15.34	-	-	-	_
Calcite	-	-	2.5	75.20	79.11
Dolomite	-	-	95.23	14.69	-
Quartz	-	-	1.42	4.72	20.42
D ₅₀ (μm)	18.6	19.8	8.13	10.2	7.58



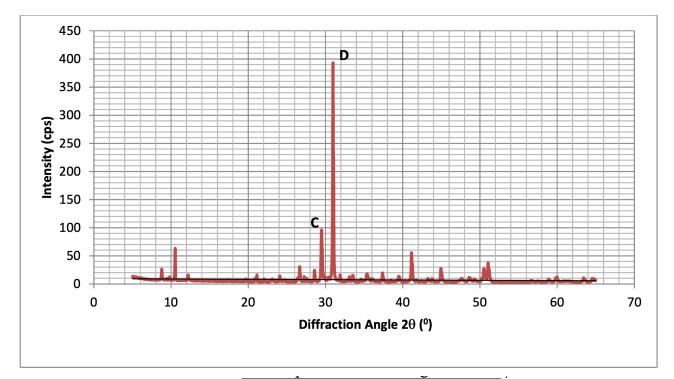
Minerology – Banswara marble



Crystaline	Wt%
Material	Rietveld
Calcite	3.07
Dolomite	95.62
Quartz	0.08
siderlite	0.57
Calcite Magnesium	0.66



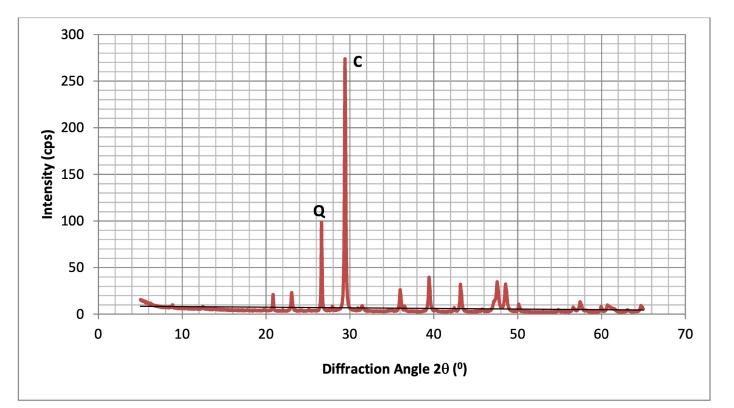
Minerology – Kishangarh marble



Crystalline material	Wt% Rietveld	
Calcite	15.48	
Dolomite	73.68	
Quartz	3.42	
siderlite	0.62	
Calcite Magnesium	6.80	



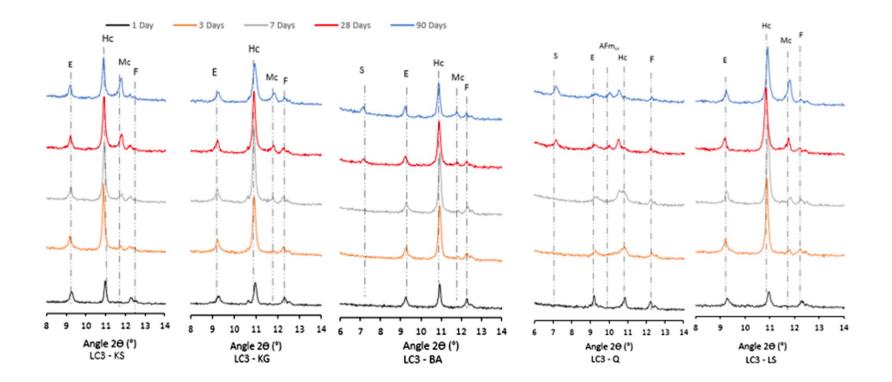
Minerology – Kota stone dust



Crystalline Material	Wt% Rietveld
Calcite	80.55
Dolomite	0.28
Quartz	16.55
Calcite Magnesium	2.64



Hydration and phase development





Hydration of clinker phases

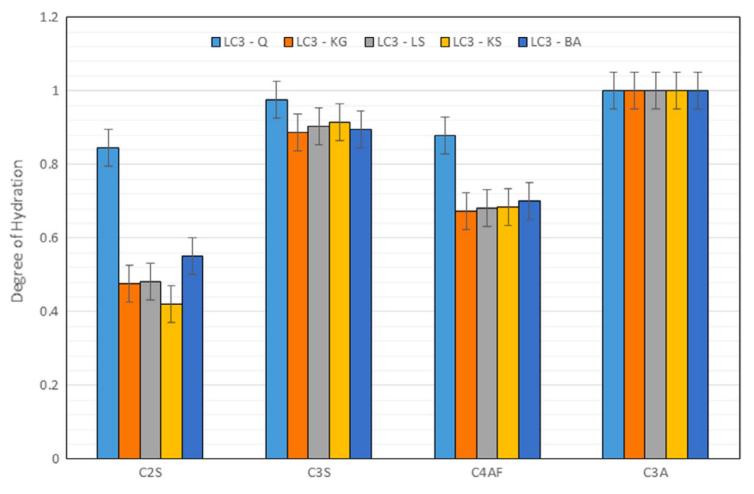
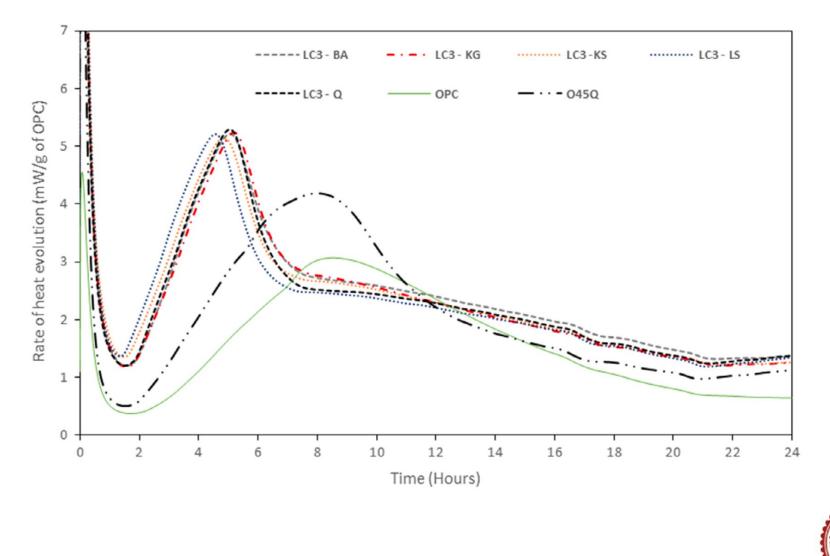


Fig. 7. The final degree of hydration of clinker phases in the blends studied at 90 days.

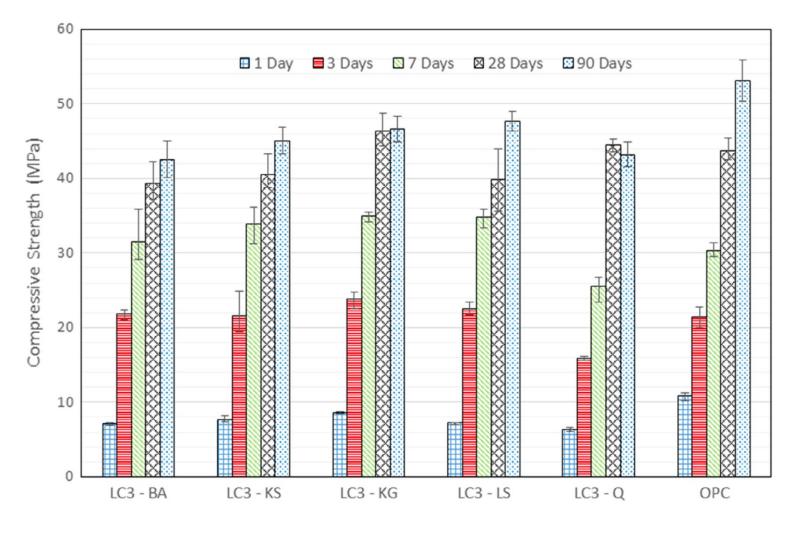


Heat of hydration



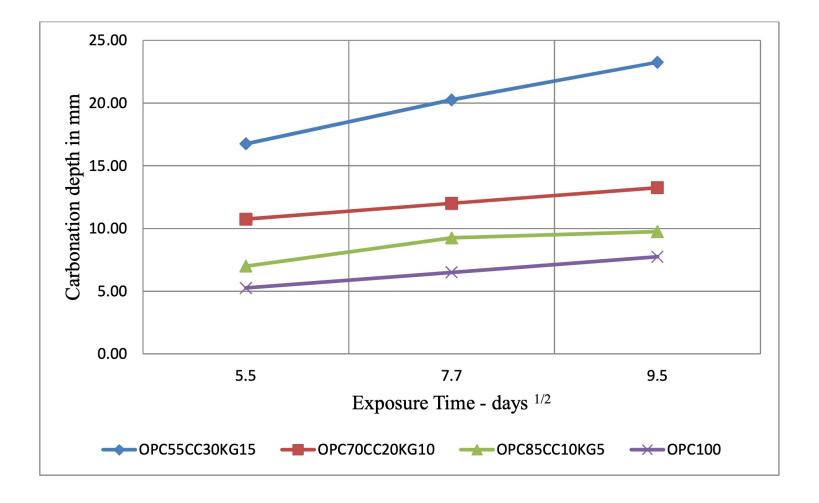
S. Bishnoi – IIT Delhi

Compressive strength





Carbonation of concrete



S. Bishnoi – IIT Delhi

Lime reactivity test (IS1727)

- 5 cm Cubes of calcium hydroxide and SCM prepared
- Water content based on flow
- Strength measured after curing at 50°C
- e.g. 4.0 MPa required for pozzolanic clay

 We find at least 6.0 MPa strength for
 reactive clays



Traditional & specialised applications



SDC building (Swiss Embassy)













S. Bishnoi – IIT Delhi

Current status

- The draft standard of LC³ to be released by BIS
- Technology is ready for commercialisation
- LC³ is commercially available in 3 countries
- Should be available in 20 to 25 more countries within a year...



Summary

- Limestone calcined clay cement
- Use of stone dusts in LC³



Thank you!



